Convenient One-Pot Synthesis of α,ω-Bis-(*para*-cyanophenyl)alkanes

Marina Yu. Lukyanova^{a,b}, Elena V. Panteleeva^{a,b}, Vitalij D. Shteingarts^{a,b}

 ^a N.N. Vorozhtsov Institute of Organic Chemistry, SB RAS, Ac.Lavrentiev Avenue, 9, Novosibirsk, 630090, Russia e-mail: <u>pantel@nioch.nsc.ru</u>
^b Novosibirsk State University, Pirogova Str., 2, Novosibirsk, 630090, Russia

 α,ω -Bis-(*para*-cyanophenyl)alkanes 1 are known as valuable precursors of biologically active compounds, monomers in the construction of macrocyclic systems and wave length converters of semi-conductor lasers. Until now there was no general method of the title compounds synthesis. Commonly used multi-stage approaches started from manifold precursors and provided total yields below 15%.¹ Recently cyanophenylalkanes 1 were found out to be the main products of the interaction of terephthalonitrile dianion as easily accessible and extremley active synthon with α,ω -dibromoalkanes in liquid ammonia. As additional components the product mixtures contained substances 2 - 5 (1÷20%, see Sheme 1). The yield of 1 varied from 60 to 90% depending on the alkyl chain length in dibromoalkane, reagents ratio and the order of their mixing. The experimrental technique allowing to synthesize 1 with the yield up to 90% have been worked out.

According to the results of testing experiments the scheme suggested for **1** formation implies initial bromine nucleophylic substitution in didromoalkane by *ipso*-carbon atom of dianion resulting in cyanophenylbromoalkan **2** followed by the formation of charge transfer complex (CTC) between **2** and dianion. Further transformations in CTC lead to cyanophenylalkanes **1** (see Sheme 2).

$$\begin{bmatrix} CN \\ CN \end{bmatrix}^{2^{-}} \xrightarrow{S_{N}} NC (CH_{2})_{n} \cdot Br \xrightarrow{(CH_{2})_{n} \cdot Br} (CH_{2})_{n} \cdot Br \xrightarrow{(CH_{2})_{n} \cdot Br} (CH_{2})_{n} \cdot Br \xrightarrow{(CH_{2})_{n} \cdot Br} (CH_{2})_{n} \cdot Br \xrightarrow{(CH_{2})_{n} \cdot CN} (CH_{2})_{n} \xrightarrow{(CH_{$$

¹ Ashley J.N., Barber H.J. JCS **1942**, 103; Sloan G.J., Vaughan W.R. J. Org. Chem. **1957**, 22, 750