Rearrangement of Caryophyllene Oxide Catalyzed by Sorbents Impregnated with Acid

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The rearrangement of caryophyllene oxide (I) in presence of alumina or silica gel impregnated with H_2SO_4 (H⁺-Al₂O₃ or H⁺-SiO₂) has been investigated.

Reaction catalyzed by H^+ - Al_2O_3 results in sum of products (II)-(XI), while the use of H^+ -SiO₂ as catalyst allows to obtain substances (XI)-(XV).



Detailed analysis of experimental data enables to conclude that H^+ -Al₂O₃-catalyzed reaction is kinetically controlled process, but compounds (**XI**)-(**XV**) are products of thermodynamically controlled reaction. Calculations of products heats of formation confirm this speculation.

Behaviour of H^+ -SiO₂ prepared from silica gel with particles less than 0.140 mm is similar to that of H^+ -Al₂O₃. This fact allows to assume the existence of dependence of catalytic activity on size of sorbent particles.